SYLLABUS - CHEM 223 - Organic Chemistry A - 1st Semester - pm lecture

Fall 2014 - LOYOLA UNIVERSITY CHICAGO (LUC)

(Lecturer, times, and sections adjusted - REVISED July 1, 2014)

2:45 - 3:35 p.m. Flanner Auditorium **Lecture**: #5716 Section: 017 Mon+Wed+Fri

Discussion: #5717 **Section:** 018 Wednesday 11:30 a.m. – 12:20 p.m. Flanner 105

#5718 **Section**: 019 Wednesday 12:35 p.m. – 1:25 p.m. Flanner 007 #5719 **Section:** 020 Wednesday 1:40 p.m. – 2:30 p.m. Flanner 007

Lecturer: Dr. C. Szpunar

> Office: Flanner Hall 200B Contact: best in person, 773-508-3128, cszpuna@luc.edu Emergency Message via Chemistry Dept. Office, 773-508-3100, fax: 773-508-3086 Office Hours: Mon and Wed: 4 – 5 p.m., Fri: 11 a.m. – 1 p.m. *** and as arranged prior

Required:

- . 1. Organic Chemistry, Wade 8th ed. (red), Prentice Hall, 2013 (ISBN 978-0-321-76841-4) *or* Wade, 7th ed. (navy blue), Prentice Hall, 2010 (ISBN 978-0-321-59231-6) *or* 6th ed. (black), Prentice Hall, 2003 (ISBN 0-13-147871-0) *or* 5th ed. (white) *or* equivalent ***NO ACCESS CODE NEEDED, NOR DESIRED, NOR SUGGESTED***
- 2. Study Guide and Solutions Manual, Wade & Simek, 8th ed. (ISBN 978-0-321-77389-0) *or* 7th ed. (ISBN 978-0-321-59871-4) *or* 6th ed. (ISBN 0-13-147882-6) *or* 5th ed., as appropriate to accompany text

Suggested / Recommended Materials:

- 1. Molecular modeling kit, Darling, Prentice-Hall, Freeman (Maruzen), Proteus, or equivalent
- 2. Spiral or bound notebook for homework problems –

Optional Materials (found helpful by some students, but hold off initially – do not purchase immediately):

- 1. Organic Chemistry as a Second Language, I, Klein (2nd edition, 2008), Wiley (ISBN 978-0-470-12929-6)
- 2. Barron's Orgo Cards: Organic Chemistry Review, Wang, Razani, Lee, Wu, and Berkowitz (ISBN 0-7641-7503-3) *or* Organic Chemistry Study Cards, R Van De Graaff, K Van De Graaff, and Prince, Morton Publishing, 2003 (ISBN 0-89582-577-5) *or* equivalent
- 3. Organic Chemistry: A Short Course, Hart, Craine, Hart, and Hadad (12th edition, 2007), w/CD-ROM, Houghton Mifflin (ISBN-10: 0-618-59073-0, ISBN-13: 978-0-618-59073-5) (Please see instructor BEFORE purchasing - only for students who may require an alternative-study approach.)

Grading (weighting below) with approximate curved grade guidelines: > 90% A; 75-90% B; 55-75% C

EXAMS – 3 – dates scheduled, announced (subject to change, although unlikely) **NO MAKE UPS 45%**

- UNEXCUSED ABSENCES merit a zero score.
- EXCUSED ABSENCES are handled on a case-by-case basis; grade weighting may be adjusted, depending on the circumstance(s); however, an excused absence MUST BE CORROBORATED and DOCUMENTED, e.g., accompanied by a note from the doctor, dentist, hospital rep, or funeral director; by a court summons, plane ticket stub, hospital release form, obituary, or other. With proper documentation, religious observance, representing the university, or personal emergency constitutes an Excused Absence.

QUIZZES – TBD – <u>unannounced</u> (during lecture or discussion or as take-home, **NO MAKE UPS**) 15% 5% HOMEWORK - assigned per topic / chapter, see Suggested Homework Assignment, below 35% FINAL EXAM – date scheduled and announced (scheduled by CAS)

^{***} Please note that because this course. Organic Chemistry, is cumulative, comprehensive, and improvementbased, and because the final exam is deemed a culminating measure of a student's progress, any student meriting an F on the final exam may achieve a recorded course grade no higher than D, despite total points; a final-exam score of D may merit a course grade no higher than C, despite total points; and a final-exam score of C may merit a course grade no higher than B, despite student's standing otherwise (i.e., despite total points.)

Course Objective: To guide, encourage, and foster the learning and understanding of Organic Chemistry – nomenclature, structures, properties, mechanisms, syntheses, and spectroscopy – by the individual student, helping him/her to connect, extrapolate, integrate, and apply the many different aspects learned.

Student Outcomes: If successful, the student will learn how to ...

- 1. identify the various classes of organic compounds, their methods of preparation, and typical reactions.
- 2. name and draw specific organic compounds.
- 3. postulate a logical reaction mechanism for simple organic reactions.
- 4. discriminate amongst relative stabilities of reaction intermediates.
- 5. plan and write out multi-step syntheses using known reagents / conditions to transform functional groups.
- 6. prepare for basic purification/separation techniques of organic compounds required in the laboratory.
- 7. analyze and interpret data from various instruments used in separating and identifying organic compounds: IR, NMR, and UV-vis spectrophotometers and mass spectrograph.

Lecture and Discussion – Attendance and Attention: Important and required. Feel free to bring your books and modeling kit to class. Better yet, use them. Prepare for lecture by prior scanning of new material. Come prepared for discussion; be ready to ask questions on assigned homework or yet-unassimilated lecture material.

Cell Phones: Please be courteous and respectful of others. Silent mode during lecture and discussion. **Not allowed in sight or within hearing during exams, subject to confiscation.** NO phone conversations in lecture hall or in discussion class – before class, during class, after class – AT ANY TIME! NO texting – before class, during class, after class – AT ANY TIME! If you must talk or text, take it outside!!!

Academic Honesty: Essential, expected, and enforced. Dishonesty dictates consequences which may include: (1) notification of Chemistry Department Chair, student's Department Chair, and CAS Dean, (2) documentation in the student's official university record, and (3) dismissal from the university. Immediate consequences will include a **ZERO** on any item in question (quiz or exam). Please refer to the LUC Undergraduate Handbook on policies or the CAS website: http://ww.luc.edu/cas/pdfs/CAS Academic Integrity Statement December 07.pdf).

Study Strategies and Suggestions: One may approach the study of Organic Chemistry in a manner similar to tackling a new foreign language. Its study will provide a basis to understanding future material – *building constantly, incessantly, and relentlessly* on the structural and mechanistic information presented previously and, hopefully, acquired by the student. Over two semesters, the course will cover: bonding, functional groups, families of aliphatic and aromatic compounds, nomenclature, structures, stereochemistry, reaction mechanisms, multi-step syntheses, and spectroscopic techniques. Because the course is cumulative and builds heavily on prior material, the best plan is to study Organic Chemistry regularly, every day, similar to practicing the piano. Collaboration on homework problems is encouraged, especially in a timely fashion. Experience dictates that positive outcomes (for exam and course grades) are directly proportional to working and understanding the assigned problems on a regular basis, i.e., applying the concepts learned to non-generic situations.

Typically, Organic Chemistry is not efficiently self-taught. Overnight cramming will probably not produce success! The student should quickly read the chapter/segment to be covered BEFORE lecture to improve lecture comprehension. After lecture, careful detailed re-reading of the chapter/segment and focused working of the assigned problems are appropriate, necessary, essential, and expected. In addition to student's participation in lecture, discussion, reading, and homework, joining and contributing to a study group is strongly encouraged.

If anticipating a passing grade of **C**, the minimal time per week devoted to Organic Chemistry is estimated at 4 hr for lecture and discussion, 4-10 hr for reading, and 4-10 hr for homework.

Homework Package: *** due Wednesday, Nov. 12, 2014 by 4 p.m. ***

Hand in – directly to the lecturer, NOT to the departmental office – at least 15 completed problems (or parts of multiple problems) per chapter, **Chapters 1–10**. (Underlined problems may be particularly insightful or anticipatory.) If not using a notebook, include your name and the date worked on **each** page. Clearly mark the chapter and problem number for full credit. Be sure to **staple** packets in chapter order ****and/or**** **flag each individual chapter in packet/notebook** (to be returned to the student). **Deliver at lecture time or to Flanner 200B.**

Suggested Homework Assignment (for Wade's 8th edition):

Chap 14: 4, 6, 8-10, 12, 14-15, (18 - m

Chap 13: 2-11, 14-15, 22-25, 32-36, 38-44

Chap 14: TBD by student

Chap 13: 2-11, 13-15, 22-25, 27, 32-36, 38-40, 42-44, 47-48 Chap 14: 4, 6, 8-10, 12, 14-15, (18 - mustard gas, bleach), 20, 22, 25, 27-33, (34), 39

(for Wade's 7th edition):

Chap 1: 2-4, 6-10, 14-15, 17-18, 20-21, 23, 25, 27, 29, 31-32, 34-36, <u>39</u>-40, 42-45, <u>54</u> Chap 2: 3-5, 7-11, 15-23, 35-36, 39-42, 44 (note solutions manual p.38 – functionality) Chap 3: 1-5, 9-10, 14-17, 20-21, 24-25, 29-30, 32-35, 37, 39, 44, 46 Chap 4: 1-2, 4, 7-13, 18-22, 26, 28-31, 34-39, 42, <u>45</u>-46, 50 Chap 5: 1-7, 15-22, 25-31 Chap 6: 1-3, 5-7, 9-12, 14-16, 19-20, 22-24, 30-38, 40-45, 53, 56 Chap 7: 1-2, 4-8, 11, 13, 15, 19, 21, 23-25, 30-36, 38, 40-46 Chap 8: 1-2, 4-15, 17-19, 21-22, 27-29, 32, 34-37, 46-47, 49, 63-<u>64</u> (N.B. solutions manual p.168 – addition + stereochemistry) Chap 9: 1-2, 5-8, 12, 18-19, 23-29, 33-34, 36-37 Chap 10: 1-4, 7-10, 13-20, 23-26, 30-33, 37-39, <u>49</u>, 51 Chap 11: 1-2, 5-6, 9-13, 21-22, 26, 31, 34, 39-45 Chap 12: 2-7, 11-12, 14-16, (17-fragmtn), 24-25

(for Wade 6th edition):

Chap 14: TBD

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Chap 1: 2-4, 7-11, 13-15, 17-18, 20, 23, 25, 27, 29, 32, 34-36, 39-40, 42, 44-45, 47
Chap 2: 5, 7-11, 15-23, 35-36, 39-42, 44
Chap 3: 1-5, 9-10, 14-15, 20-21, 29-35, 37, 39, 44, 46
Chap 4: 1-2, 4-13, 18-22, 26, 28-31, 34-39, 42, 45-46, 50
Chap 5: 1-3, 5-8, 16-22, 25-30
Chap 6: 1-7, 11-12, 14-16, 19-20, 22-24, 31-37, 44-45, 53, 56
Chap 7: 1-2, 4-5, 7-8, 13, 15, 19, 21, 23, 25, 34-35, 38, 40, 42-45
Chap 8: 1-2, 4-6, 8-15, 17-19, 27-28, 32, 34-37, 47, 49, 63-64
Chap 9: 1, 2, 5-8, 12, 18-19, 23-24, 33-34
Chap 10: 3, 7, 8-10, 13-15, 17-20, 24-26, 37-39, 49
Chap 11: 2, 5-6, 9-11, 13, 22, 31, 34, 41-45
Chap 12: 2-12, 14-19, 23, 25
Chap 13: 2-36, 38-45
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Lecture Outline (subject to change) - Fall 2014

<u>Week</u>	<u>Date</u>	Chap-Lecti	<u>Ire Topic</u>		*** <u>EVENT</u> ***
1	Aug 25 27		Intro: Lewis structures	, bonding, resonanc	e, acid-base, nomenclature
2	29 Sept 1	<u>2-1</u> ***	Structure and propertie		LABOR DAY - HOLIDAY ***
3	3 5 Sept 8	3-1	Alkanes, cycloalkanes,	bicyclics	
4	10 12 Sept 15	4-2	Chemical rxns - free ra	dical halogenation,	kinetics, intermediates
	17 19	5-1 <u>5-2</u>	Stereochemistry – chira	ality, isomers	
5	Sept 22 24 26	***	******* WEDNE Alkyl halides, nucleoph		EXAM I (Chapters 1-4) elimination
6	Sept 29 Oct 1	6-3			
7	Oct 6/7	7 *** 7-1	**************************************	flon/Tues ********	***** MIDTERM BREAK ***
8	Oct 13	7-3 8-1	Alkenes – rxns		
9	Oct 20	***	****** MONL	DAY *******	EXAM II (Chapters 5-7)
40	24		Alkynes		
10	Oct 27 29 31	9-2 9-3 <u>10-1</u>	Alcohols ***** Fr	<i>iday</i> ***** (last day	to withdraw with a W) *****
11	Nov 3 5 7		Alcohols – rxns		
12	Nov 10 12	11-2 11-3	*** W ed	-	ssignment due for grade ***
13	Nov 17 19	12-2 12-3	Spectroscopy – IR and		
	21		****** FRIDA	4Y ******** <i>E</i>	XAM III (Chapters 8-11)
14 15	Nov 24 26-2 Dec 1		Spectroscopy – NMR ******* Wed-Fi	ri ***** THANKS	GIVING DAY - HOLIDAY ***
16	3 5	13-3	•		er own, if time does not permit lecture coverage) p.m., Flanner Auditorium